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GLC-Mass Spectral Analysis of Fungal Metabolites

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Abstract □ Four metabolites, hispidin, bisnoryangonin, muscimole, and ibotenic acid, from potentially psychoactive mushrooms were analyzed by GLC-mass spectrometry as their trimethylsilyl derivatives. This method was applied to the first two compounds in *Gymnopilus punctifolius* (Peck) Singer and to the last two compounds in *Amanita pantherina* (Fr.) Secr.

Keyphrases □ Hispidin—GLC-mass spectral analysis in mushrooms □ Bisnoryangonin—GLC-mass spectral analysis in mushrooms □ Buscimole—GLC-mass spectral analysis in mushrooms □ Botenic acid—GLC-mass spectral analysis in mushrooms □ GLC-mass spectrometry—analyses, hispidin, bisnoryangonin, muscimole, and ibotenic acid in mushrooms □ Gymnopilus punctifolius—GLC-mass spectral analysis of various components □ Amanita pantherina—GLC-mass spectral analysis of various components

A previous report described the GLC-mass spectral analysis of the trimethylsilyl derivatives of psilocin and psilocybin (1). As part of continuing studies of the chemical constituents of higher fungi, this report describes the GLC analysis of the trimethylsilyl derivatives of four such compounds and their mass spectral features. Bisnoryangonin (Ia), hispidin (IIa), muscimole (IIIa), and ibotenic acid (IVa) occur in various genera of the Hymenomycetes.

BACKGROUND

The styrylpyrone derivatives Ia and IIa were analyzed by TLC and IR, UV, and mass spectrometry (2, 3). Compound IIa was first reported (4-6) in certain species of Polyporaceous fungi. One or both of these compounds have been detected in various species of *Gymnopilus* Karsten (Cortinariaceae), *Pholiota* (Fr.) Kummer [including *Flammula* (Fr.) Kummer] (Strophariaceae or Cortinariaceae), and *Naematoloma* Karsten [= *Hypholoma* (Fr.) Kummer] (Strophariaceae) (2, 7-10). These and related compounds may be of considerable use in biochemical systematics (10-13).

The historical literature of Japan suggests the ethnomycological importance of G. spectabilis (Fr.) Smith¹ [= Pholiota spectabilis (Fr.) Quel.] (15, 16), but only occasional incidents of intoxication due to this species have been reported recently (17, 18). The presence of the related styryl-pyrone yangonin (V) in the rhizomes of kava-kava (*Piper methysticum* Forst.) (19), the narcotic pepper used in the South Pacific, lends support to the possibility that this class of compounds might be psychoactive. However, limited pharmacological studies of the pure kava pyrones have not accounted for the activity of the whole plant (20).

Compounds IIIa and IVa were analyzed previously by paper chromatography (21-23) and paper electrophoresis (21, 24). Instrumental



analysis of the pure compounds was performed by IR (21) and mass spectrometry (25). These compounds appear to be confined to a few species of the genus Amanita Pers. ex Hook., principally A. muscaria (L. ex Fr.) Hooker (21–23, 26), A. pantherina (Fr.) Secr. (23, 26), and A. cothurnata Atk. (26). The ethnobotanical importance of A. muscaria has been established (27). Although these mushrooms are the cause of numerous intoxications, both accidental and intentional (28), pure compounds IIIa and IVa have received limited pharmacological study (29–31).

The previously described methods have been adequate for the qualitative analysis of these compounds. However, quantitative analysis by these methods suffers from several disadvantages, especially for the highly polar, zwitterionic IIIa and IVa.

The facile decarboxylation of IVa to IIIa during extraction, analytical, and isolation procedures (23, 26) leads to an inaccurate estimation of individual components present in the mushroom. Occasionally, enrichment of fungal extracts by ion-exchange chromatography is necessary before analysis can proceed (26). Use of colorimetric reagents to visualize spots on a chromatogram can often be misleading, especially in quanti-

 Table I—Chromatography Conditions, Retention Times *, and

 Percent of Trimethylsilyl Derivatives

	Ib	IIb	ΠIb	IVb	IVc
SE-30	250°, <i>B</i> . 85 min	250°, R. 15 min	$150-250^{\circ}$, <i>R</i> , 2.1 min	150250°, B. 5.4 min	150–250°, B. 3.7 min
OV-101	<i>It</i> 0.5 IIIII		$100-200^{\circ}$, R, 60 min	$100-200^{\circ}$, B, 13.1 min	$100-200^{\circ}$, <i>R</i> , 10.8 min
Percent ^b present in fungal extracts	0.44 (0.42) ^c	$0.25 \\ (0.21)^c$	0.046	0.002	

^a Relative to the solvent front. ^b Based on dry weight. ^c By isolation.

¹ This species might be identical with G. junonious (Fr.) Orton (14).

tative estimation. Most of these difficulties can be overcome utilizing GLC of the neutral, stable, and easily prepared derivatives Ib, IIb, IIIb, and IVb.

EXPERIMENTAL²

Muscimole monohydrate was prepared according to published procedures (32, 33). The melting point and IR and mass spectra were in agreement with those reported (21, 25).

Pileus tissue of air-dried G. punctifolius (Peck) Singer³ (2) (89 mg) was ground to a powder with sand and transferred to a screw-capped tube. Methanol (5 ml) was added, and the tube was shaken at room temperature for 20 hr. The contents were filtered, and the collected solids were washed with two 5-ml portions of methanol. Then the combined filtrate and washings were concentrated to 5.0 ml, and 0.5 ml of this solution was transferred to a 1.0-ml vial4.

The solvent was removed by evaporation in a nitrogen stream, and the residue was dried in vacuo. Bis(trimethylsilyl)trifluoroacetamide⁵ (25 μ l) was added, and the vial was closed with a septum-lined seal⁶ and heated for 30 min at 140°. Aliquots of 1 μ l were used for subsequent analyses.

Pileus tissue of freeze-dried A. pantherina (Fr.) Secr.⁷ (1.56 g) was ground as already described and extracted with 30 ml of 10% aqueous methanol at 4° for 8 hr. The mixture was filtered, and the volume of the filtrate was adjusted to 30.0 ml. An aliquot of this solution (0.5 ml) was transferred to a reaction vial and treated as already described.

Quantitation was carried out based on the external standard technique used previously (1). Weighed samples of each of the four pure standards were mixed with bis(trimethylsilyl)trifluoroacetamide in reaction vials so that the final concentrations were 0.25% (w/v). The sealed vials were heated at 140°. Aliquots were removed every 15 min and analyzed by GLC for a total of 2 hr.

In each case, the reaction solutions attained homogeneity within 15-30 min; analyses of aliquots removed after 30 min showed no further increase in peak areas. When injection of equal volumes of each sample was repeated, reproducibility of detector response (as measured by peak areas) was within $\pm 2\%$. Linearity of detector response was verified by injecting successively diluted volumes of each standard solution and comparing the peak areas obtained with the known concentrations of solutes.

Carpophores of G. punctifolius³ (1.911 g) were ground and extracted by stirring with methanol (100 ml) for 20 hr. The solids were collected by filtration, and the filtrate was concentrated in vacuo to 10.0 ml. A second 100-ml extract of the filtered solids was shown to be devoid of Ia and IIa by TLC. Compounds Ia and IIa were isolated from the extract by preparative TLC, using 25% methanol in chloroform as the developing solvent. The separated bands (Ia, R_f 0.58; and IIa, R_f 0.44) were removed from the plates, and the compounds were eluted with 40% methanol in ethyl acetate.

The residue of Ia, obtained by concentration of the eluate, was crystallized from methanol-ether to afford 8.0 mg (0.42%), mp 242-244° dec. [lit. (2) mp 244-246° dec.]. The residue of IIa was precipitated from

³ G. punctifolius (Peck) Singer was collected in Grays Harbor County, Washington.
 ³ G. punctifolius (Peck) Singer was collected in Grays Harbor County, Wash., in October 1975. This material is on deposit at the University of Washington Her-barium, Seattle, Wash., as LESLIE 2663.
 ⁴ Hewlett-Packard 5080-8712.
 ⁵ Devid Devided Doc.

⁶ Hewlett-Packard 5080-8713.

⁷ A. pantherina (Fr.) Secr. was collected in Thurston County, Wash., in October 1975 and is on deposit at the University of Washington Herbarium, Seattle, Wash., as LESLIE 2629.

Table II—Mass	Spectral	Parameters ⁴	I for	Trimethylsilyl
Derivatives	-			• •

$\begin{array}{cccccccccccccccccccccccccccccccccccc$	Ib	IIb	IIIb	IVb	IVc
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	372 (7%)) 464 (12%) 2) $(M + 2)$	244 (2%)	360 (2%)	272 (1%)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	375 (189 (M + 1	%) $463 (22\%)$ (M + 1)	243 (8%) (M - 15)	359 (6%) (M - 15)	259 (2%)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	874 (519 (M +)	%) $462 (50\%)$ (M +)	169 (5%)	332 (2%)	258 (4%)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	361 (2%) 449 (7%)	156 (2%)	331 (6%) (M - 43)	257 (20%)
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	360 (6% 359 (199) $448 (13\%)$ %) $447 (24\%)$	147 (2%) 146 (3%)	259 (7%) 258 (9%)	156 (2%) 147 (3%)
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	(M - 1) 347 (6%	(15) $(M - 15)$ (15) $434 (7\%)$	144 (2%)	257 (45%)	121 (5%)
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	346 (199	(M - 28) %) 419 (5%)	128 (4%)	(M - 117) 218 (3%))
$\begin{array}{c} (M-56) \\ (M-56$	(M - 2)	28)) 407 (3%)	114 (3%)	169 (2%)	100 (4%)
$ \begin{array}{c} (M-56) \\ 317 & (2\%) & 359 & (4\%) & 100 & (3\%) & 147 & (15\%) & 89 & (5\%) \\ 219 & (3\%) & 346 & (7\%) & 99 & (2\%) & 133 & (2\%) & 75 & (15\%) \\ 99 & (4\%) & 245 & (3\%) & 89 & (1\%) & 130 & (4\%) & 74 & (11\%) \\ 75 & (13\%) & 229 & (4\%) & 75 & (10\%) & 103 & (2\%) & 73 & (100\%) \\ 74 & (9\%) & 219 & (4\%) & 74 & (11\%) & 100 & (5\%) & 45 & (17\%) \\ 73 & (100\%) & 191 & (9\%) & 73 & (100\%) & 99 & (2\%) \\ 69 & (4\%) & 183 & (5\%) & 59 & (8\%) & 75 & (11\%) \\ 45 & (15\%) & 147 & (10\%) & 45 & (18\%) & 74 & (9\%) \\ & & 143 & (5\%) & 44 & (3\%) & 73 & (100\%) \\ & & & 99 & (6\%) & 43 & (6\%) & 59 & (3\%) \\ \end{array} $	318 (189	(M - 56) %) 406 (8%)	102 (10%)	148 (3%)	99 (3%)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	(M - 317)	56) 359 (4%)	100 (3%)	147 (15%)	89 (5%)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	219 (3%) 346 (7%)	99 (2%)	133 (2%)	75 (15%)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	99 (4%) 245 (3%)	89 (1%)	130 (4%)	74 (11%)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	75 (139	%) 229 (4%)	75 (10%)	103 (2%)	73 (100%)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	74 (9%) 219 (4%)	74 (11%)	100 (5%)	45 (17%)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	73 (100)%) 191 (9%)	73 (100%)	99 (2%)	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	69 (4%	183(5%)	59 (8%)	75 (11%)	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	45 (15%	%) 147 (10%) 142 (5%)	40 (18%)	74 (9%)	
33 (0%) 43 (0%) 33 (3%)		143 (3%)	44 (3%) 12 (6%)	- 73 (100%) - 50 (9%)	
75 (33%) 45 (10%)		75 (33%)	40 (0%)	45 (10%)	
74 (35%)		74 (35%)		40 (10%)	
73 (100%)		73 (100%)			
69 (5%)		69 (5%)			
45 (48%)		45 (48%)			

^a m/e (relative intensity).

ethanol by addition of ethyl acetate, and the precipitate was isolated by centrifugation. After drying in vacuo, IIa weighed 4.0 mg (0.21%), mp 228° [lit. (4) mp 256°]. The compounds were chromatographically homogeneous and indistinguishable from reference standards.

RESULTS AND DISCUSSION

The derivatives Ib and IIb were widely separated under the isothermal conditions described. While IIb was completely resolved from other components in the natural extract, some overlap with an adjacent peak was noted for Ib. However, a mass spectrum comparable to that of the standard was obtained. The resolution of Ib was not improved by changes in chromatographic conditions or stationary phases. The isoxazoles IIIb and IVb were well resolved under either set of conditions.

The concentrations of Ia and IIa in the collection of G. punctifolius used in this study were lower than those reported for this species previously (2). The quantities of IIIa and IVa in A. pantherina also appeared to be low compared to earlier findings (23). Both previous investigations were based on analysis by spot-intensity techniques on paper and thinlayer chromatograms. However, these methods are only semiquantitative (34). The isolated yields of IVa recently reported (26) from A. pantherina and of Ia and IIa from G. punctifolius reported here are compatible with data obtained by GLC (Table I).

Several factors could affect the quantities of Ia, IIa, IIIa, and IVa found in these organisms, including intraspecies variation based on season, year, and environment of collections and differences in handling and storage of specimens. With respect to the former, marked variation in isoxazole content was reported in collections of A. muscaria (35). The quantity of hispidin in Polyporus hispidus (Bull.) Fr. decreased with the increasing age of the carpophore (5). It was suggested that this decrease was due to the elaboration of hispidin into polymeric lignans; recently, polymeric hispidin derivatives were detected in several species containing Ia and IIa (10-13).

The mass spectra of the styrylpyrone trimethylsilyl derivatives (Table II) were characterized by intense molecular ion signals (m/e 462 and 374 for IIb and Ib, respectively). This result is rather unusual since trimethylsilyl ethers undergo facile loss of a methyl group (36). These M - 15 peaks were also present in the spectra of *Ib* and *IIb*. The base peak in each spectrum at m/e 73 due to the trimethylsilyl ion is a common occurrence in these derivatives (36).

² GLC was carried out with a Hewlett-Packard model 402 instrument with GLC was carried out with a newett-rackard model 402 instrument with flame-ionization detectors. The columns used were: (a) a 0.75-m \times 2.8-mm i.d. glass U-tube with 3% OV-101 on 100-120-mesh Gas Chrom Q with temperature pro-grammed from 100 to 200° at 7.5°/min, and (b) a 1.2-m \times 2.8-mm i.d. glass U-tube with 1.5% SE-30 on 100-120-mesh Gas Chrom W with temperature programmed from 150 to 250° at 7.5°/min or isothermally at 250°. The injector block temperature use 250° the detector uses at 280° and the chart energy use 0.64 cm/min with belium was 250°, the detector was at 280°, and the chart speed was 0.64 cm/min with helium carrier gas at 50 ml/min. GLC-mass spectrometa

⁻mass spectrometry was performed with a Finnigan 9500 gas chromatograph coupled to a Finnigan 3100 D quadrupole mass spectrometer through a single-stage glass jet separator. Data were acquired *via* a System/250 digital computer system. The mass spectrometer conditions were: interface temperature, 225°; transfer line, 1759 ; manifold temperature, 100°; and ion source potential, 70 ev. Chromatography conditions were the same as those used for GLC, except that the helium flow rate

was 20 ml/min. Analytical TLC was carried out on 0.25-mm layers of silica gel GF on glass plates. Preparative TLC was called out of 0.25 min layers of since get GF of glass plates. Preparative TLC was performed with 1-m $\times 20$ cm glass plates coated with 0.75 mm layers of silica gel GF. The solvent system in both cases was 25% methanol in chloroform

Standards of hispidin and bisnoryangonin were obtained from Dr. Lynn R. Brady, Department of Pharmaceutical Sciences, School of Pharmacy, University of Washington, Seattle, Wash. Natural ibotenic acid was obtained from Dr. William

⁵ Regisil, Regis Chemical Co.



Scheme I

The fragmentation processes and rearrangements in the mass spectra of isoxazoles are complex (37-39). Scheme I shows several pathways proposed for the mass spectrum of IIIb. The rearrangement isoxazole --azirine \rightarrow oxazole $(A \rightarrow B \rightarrow C)$ (37) can lead to parallel fragmentation processes. Loss of methyl from a silyl group from any of the rearranged parent ions accounts for the m/e 243 signal. This process occurs so readily that the parent ion (m/e 258) is not evident. Production of trimethylsilylmethylimminium ion $(m/e \ 102)$ followed by loss of carbon monoxide can lead to ions D (m/e 156) and E (m/e 128). Also, loss of trimethylsilylamine can lead to ions such as $F(m/e \ 169)$, which can undergo further rearrangement. Similar fragmentations of ion C could also account for the signals at m/e 156 and 169.

The mass spectrum of IVb exhibited all typical fragmentations of trimethylsilylated α -amino acids (40), including M - 15 (m/e 359), M - 43 (331), M - 117 (257) due to loss of carbonyltrimethylsilyloxy, and m/e 218 due to loss of the amino acid side chain (36). The parent peak (m/e 374) was not present. Shorter reaction times or lower reaction temperatures resulted in the presence of variable amounts of a partially derivatized product, presumably IVc. The production of more than one trimethylsilyl derivative when various silylating reagents are used is not an uncommon occurrence with α -amino acids (40). However, quantitative derivatization giving only IVb occurred under the described conditions (see Experimental). Loss of carbon dioxide and a proton from IVc with retention of the trimethylsilyl group would account for the m/e 257 signal. This compound also exhibited fragments similar to those discussed for IIIb.

The decarboxylation of IVa to IIIa (41) did not occur during silylation of IVa since IIIb was not present in gas chromatograms of the reaction solutions. GLC offers accurate quantitative analysis of Ia, IIa, IIIa, and IVa directly from initial extracts. Sensitivity is in the 10^{-8} - 10^{-7} -g range with analyses routinely run on 10-20 mg of dried fungal tissue. The strong signals for the parent ions of Ib and IIb and the combination of characteristic fragmentations for IIIb and IVb allow positive identification of these substances.

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